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Publisher *Taylor & Francis*

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Organic Preparations and Procedures International

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t902189982>

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To cite this Article Lee, Hongmee and Harvey, Ronald G.(1988) 'SYNTHESES OF OXYGENATED DERIVATIVES OF 3-METHYLCHOLANTHRENE', *Organic Preparations and Procedures International*, 20: 2, 123 – 128

To link to this Article: DOI: 10.1080/00304948809355799

URL: <http://dx.doi.org/10.1080/00304948809355799>

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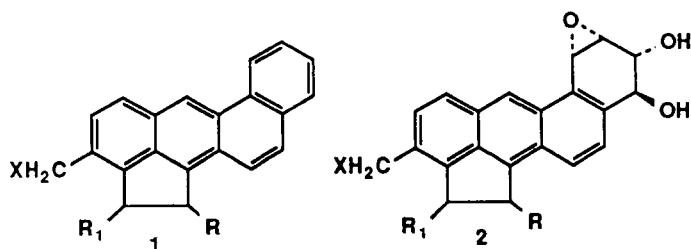
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SYNTHESES OF OXYGENATED DERIVATIVES OF 3-METHYLCHOLANTHRENE

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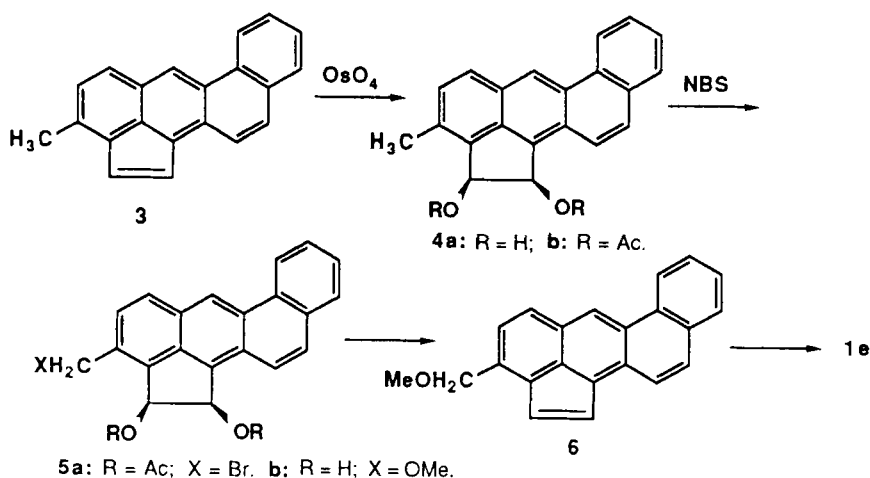
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Carcinogenic polycyclic aromatic hydrocarbons are believed to be activated enzymatically to reactive diol epoxide metabolites which bind covalently to DNA.¹⁻³ The potent carcinogen 3-methylcholanthrene (3-MC) (1a) appears exceptional in that its bay region diol epoxide derivative (2a)⁴ accounts for only a small fraction of the DNA-bound adducts in mammalian cells.⁵ Metabolism studies indicate that the principal active species is a triol epoxide containing an additional hydroxyl group in the 1-, 2-, or 3-positions (2b-d).⁵⁻⁹ Further elucidation of the mechanism of metabolic activation of 3-MC is dependent upon the synthetic accessibility of the appropriate oxygenated derivatives of 3-MC. We now report improved syntheses of 1- and 2-hydroxy-3-MC (1b,1c) and a convenient synthesis of the methyl ether of 3-hydroxymethylcholanthrene (1e).



a: R = R₁ = X = H. b: R = OH; R₁ = X = H. c: R = X = H; R₁ = OH.
d: R = R₁ = H; X = OH. e: R = R₁ = H; X = OMe.

Synthesis of 1e was based on 3-MC¹⁰. The greater reactivity of the methylene groups of 3-MC in relation to the methyl function necessitated introduction of blocking groups into these positions. This was accomplished by conversion of 3-MC to cis-1,2-diacetoxy-3-MC (4b) via dehydrogenation with DDQ



to 3-methylcholanthrylene (3), followed by *cis*-dihydroxylation with OsO_4 , and acetylation. Bromination of 4b with NBS catalyzed by benzoyl peroxide afforded smoothly the 3-bromomethyl diacetate derivative 5a. Treatment of 5a with sodium methoxide in methanol resulted in displacement accompanied by concurrent deacetylation to yield *cis*-1,2-dihydroxy-3-methoxy-methylcholanthrene (5b). Deoxygenation of 5b with triphenyl phosphite, imidazole and iodine¹¹ furnished 3-methoxymethylcholanthrylene (6) which was hydrogenated over a Pd/charcoal catalyst to yield 1e. Attempted demethylation of 1e with various reagents (BBr_3 , HCl , Ph_3CBF_4) to generate the free alcohol were unsuccessful.¹²

Oxidation of 1a with lead tetraacetate in benzene at 0°C gave pure 1b acetate virtually quantitatively, a significant improvement over the yield (46%) obtained from analogous reaction in glacial acetic acid.¹³ Treatment of 1b acetate with KOH in methanol provided pure 1b. Oxidation of 1a with DDQ in aqueous acetic acid gave 3-methylcholanthren-1-one (68%) which underwent reduction with NaBH_4 in THF-methanol to also yield 1b (98%).

2-Hydroxy-3-MC (1c) was conveniently prepared from 3-methylcholanthren-2-one obtained by acidic dehydration of the dihydrodiol 4a.¹⁴ Reduction of the ketone with NaBH_4 furnished pure 1c (98%).

The compounds **1b**, **1c**, and **1e** were employed in metabolism studies which have led to identification of **2d** as the principal DNA-bound metabolite of 3-MC in mouse cells.⁵ Failure to demethylate **1e** was not a problem for the biological studies, since the hydroxylated metabolites could be readily methylated for comparison.

EXPERIMENTAL SECTION

Materials and Methods. 3-Methylcholanthrene was prepared by the method described.¹⁰ 2,3-Dichloro-5,6-dicyano-1,4-benzoquinone (DDQ) was recrystallized from benzene. N-Bromosuccinimide (NBS) was crystallized from water prior to use. Tetrahydrofuran (THF) was freshly distilled from LiAlH₄. The NMR spectra were recorded on a Varian EM-360 60 MHz spectrometer with tetramethylsilane as an internal standard in CDCl₃. Melting points are uncorrected.

3-Methylcholanthrylene (3).- A solution of **1a** (4.43 g, 16.4 mmol) and DDQ (4.10 g, 18 mmol) in dry benzene (400 ml) was stirred at room temperature for 1 hr. The solution was passed through a column of neutral alumina eluted with benzene to afford **3** (3.46 g, 79%) as a yellow solid: mp 196-197°C (lit.¹⁴ 194°C); NMR δ 2.65 (s, 1, CH₃), 7.1-8.2 (m, 9, Ar and H_{1,2}), 8.5 (m, 1, H₇), 9.1 (s, 1, H₆).

cis-1,2-Diacetoxy-3-methylcholanthrene (4b).- To a suspension of **3** (2.12 g, 7.9 mmol) in 10 ml of pyridine was added a solution of OsO₄ (2 g, 7.9 mmol) in 50 ml of pyridine. The mixture was stirred at room temperature for 4 days. A solution of NaHSO₃ (3.6 g) in 60 ml of H₂O and 40 ml of pyridine was added. The precipitate of **cis-1,2-dihydroxy-3-methylcholanthrene** (2.06 g) was filtered and dried. A suspension of the diol in pyridine (50 ml) and acetic anhydride (70 ml) was stirred at ambient temperature for 4 days. The mixture was poured into ice water. The precipitate was dissolved in benzene and chromatographed on a column of Florisil. Elution with EtOAc/CH₂Cl₂ (1:1) afforded **4b** (1.5 g, 50%) as a white solid: mp 210-212°C; NMR δ (s, 6, COCH₃), 2.5 (s, 3, CH₃), 6.85 (d, 1, H₁, J_{1,2} = 6 Hz), 7.3 (d, 1, H₂), 7.4-8.1 (m, 7, Ar), 8.6 (m, 1, H₇), 9.1 (s, 1, H₆); MS(EI) calcd for C₂₅H₂₀O₄, m/e 384, found 384.

3-Bromomethyl-cis-1,2-diacetoxycholanthrene (5a).- To a solution of **4b** (386 mg, 1 mmol) in CCl₄ (25 ml) were added a few crystals of benzoyl peroxide and NBS (200

mg, 1.1 mmol). The solution under N_2 was refluxed under the irradiation of a sun lamp for 1 hr. The resulting suspension was dissolved in CH_2Cl_2 and washed with H_2O three times. Evaporation of the solvent afforded **5a** (460 mg, 99%) as a white solid virtually pure by NMR : mp $> 230^\circ C$; NMR δ 2.2 (s, 3, $COCH_3$), 2.3 (s, 3, $COCH_3$), 4.65 (s, 2, CH_2), 7.2 (q, 2, methine), 7.5-8.2 (m, 7, Ar), 8.8 (d of d, 1, H_7), 9.2 (s, 1, H_6); MS(EI) calcd for $C_{25}H_{19}O_4Br$, m/e 462, found 462, 464 (p+2).

cis-1,2-Dihydroxy-3-methoxymethylcholanthrene (5b).- A solution of **5a** (460 mg, 1 mmol) in THF (20 ml), MeOH (50 ml) and 1 N NaOH (10 ml) was stirred at ambient temperature for 3 hr. The solution was worked up conventionally to afford **5b** (292 mg, 88%) as a yellow solid: mp $> 230^\circ C$; NMR (acetone- d_6/D_2O) δ 3.4 (s, 3, CH_3), 4.9 (s, 2, CH_2), 5.8 (q, 2, methine), 7.6-8.5 (m, 7, Ar), 8.9 (m, 1, H_7), 9.3 (s, 1, H_6).

3-Methoxymethylcholanthrylene (6).- To a solution of **5b** (76 mg, 0.23 mmol), triphenyl phosphite (572 mg, 1.84 mmol), and imidazole (126 mg, 1.84 mmol) in 20 ml of benzene was added I_2 (250 mg, 1.4 mmol). The solution was stirred for 45 min, then diluted with ether, washed with aqueous $NaHCO_3$ and $NaHSO_3$ solutions, dried over $MgSO_4$, and evaporated to dryness. The crude product was chromatographed on a Florisil column eluted with ether to yield **6** (44 mg, 65%) as a yellow solid : mp $162-163^\circ C$; NMR δ 3.4 (s, 3, CH_3), 4.85 (s, 2, CH_2), 7.2-8.2 (m, 7, Ar), 8.8 (m, 1, H_7), 9.2 (s, 1, H_6); MS(EI) calcd for $C_{22}H_{16}O$, m/e 296, found 296.

3-Methoxymethylcholanthrene (1e).- Compound **6** (47 mg, 0.24 mmol) dissolved in ethyl acetate (120 ml) and benzene (80 ml) was hydrogenated over a 10% Pd/C catalyst (25 mg) at 10 psig at ambient temperature for 20 min. The usual workup followed by chromatography on a column of Florisil afforded **1e** (29 mg, 62%) as a white solid: mp $148-149^\circ C$; NMR δ 3.3 (s, 3, CH_3), 3.5 (m, 4, $H_{1,2}$), 4.55 (s, 2, CH_2), 7.2-7.9 (m, 7, Ar), 8.7-9.5 (m, 2, $H_{6,7}$); MS(EI) calcd for $C_{22}H_{18}O$, m/e 298.135, found 298.134.

1-Hydroxy-3-methylcholanthrene (1b).- To a solution of **1a** (135 mg, 0.5 mmol) in 20 ml of benzene was added a suspension of $Pb(OAc)_4$ (244 mg, 0.55 mmol) at 0°

cover 1 hr. The reaction mixture was concentrated to 1/3 the original volume and the residue was diluted with benzene. The organic phase was washed with water, dried and the solvent was evaporated to give **1b** acetate (160 mg, 98%) which was virtually pure by NMR: mp 167.5-168.5°C (benzene) (lit.¹³ 177.5-178.5°C); NMR δ 2.2 (s, 3, COCH₃), 2.5 (s, 3, CH₃), 3.0-4.2 (m, 2, CH₂), 7.1-9.1 (m, 8, methine and Ar), 8.9 (m, 1, H₇), 9.2 (s, 1, H₆); MS(EI) calcd for C₂₃H₁₈O₂, m/e 326, found 326.

A solution of **1b** acetate (200 mg, 0.6 mmol) in 5% KOH in methanol (130 ml) was heated at reflux for 40 min. The solvent was evaporated and ice water was added to the residue. The precipitate which formed was collected and dried to provide **1b** (180 mg, 99%) which was pure by HPLC: mp 215-216°C (dec.) (lit.¹³ 214-216°C).

To a solution of **1a** (135 mg, 0.5 mmol) in acetic acid (130 ml) and H₂O (10 ml) was added DDQ (250 mg, 1.1 mmol) at 40°C. The solution was stored at 40°C for 1 hr during which time the dark green color changed to dark red. The solution was cooled, diluted with ether, and extracted with H₂O and aqueous NaOH. The organic phase was dried and the solvent was evaporated to dryness. The crude product was chromatographed on a Florisil column eluted with ether to yield 3-methylcholanthren-1-one (97 mg, 68%) as a yellow solid: mp > 250°C (lit.¹³ 262-3°C); NMR δ 2.5 (3, s, CH₃), 3.7 (2, s, CH₂), 7.2-9.2 (8, m, aromatic), 9.3 (1, s, H₆).

A solution of 3-methylcholanthren-1-one (97 mg, 0.34 mmol) and NaBH₄ (100 mg) in THF (30 ml) and MeOH (30 ml) was stirred for 1 hr at room temperature. The usual workup gave **1b** (95 mg, 98%) as a white solid which was identical with that prepared via the alternative procedure.

2-Hydroxy-3-methylcholanthrene (1c).- 3-Methylcholanthren-2-one was synthesized according to the procedure of Sims¹⁴ : mp 203-204°C (lit.¹⁴ 205°C); NMR δ 2.85 (s, 3, CH₃), 3.95 (s, 2, CH₂), 7.2-8.2 (m, 6, Ar), 8.7 (m, 1, H₇), 9.05 (s, 1, H₆).

A solution of this ketone (118 mg, 0.42 mmol) and NaBH₄ (120 mg) in THF (30 ml) and MeOH (30 ml) was stirred for 1 hr at room temperature. The usual workup

gave **1c** (115 mg, 98%) as a white solid: mp 219-220°C (EtOAc) (lit.¹⁴186°C). The alcohol was further characterized by acetylation with acetic anhydride and pyridine at room temperature overnight. After the usual workup, the **1c** acetate was crystallized from benzene-ether to give pure **1c** acetate: mp 173-174°C; NMR δ 2.2 (s, 3, COCH₃), 2.5 (s, 3, CH₃), 3.5 and 4.2 (pair of d of d, 2, CH₂), 6.85 (d of d, 1, methine), 7.3-8.1 (m, 7, Ar), 8.8 (m, 1, H₇), 8.95 (s, 1, H₆); MS(EI) calcd for C₂₃H₁₈O₂, m/e 326, found 326.

Acknowledgement.— This research was supported by grants from the National Cancer Institute (CA 36097 and CA 14459) and the American Cancer Society (BC-132).

REFERENCES

1. R. G. Harvey "Polycyclic Hydrocarbons and Carcinogenesis," ACS Symp. Series No. 283, American Chemical Society, Washington, D. C., (1985).
2. R. G. Harvey, *Acc. Chem. Res.*, **14**, 218 (1981).
3. A. Dipple, R. C. Moschel, and C. A. H. Bigger, "Chemical Carcinogens," 2nd Ed, C. E. Searle, Ed., ACS Monograph No. 182, American Chemical Society, Washington, D. C., 1984, pp 41-163.
4. S. A. Jacobs, C. Cortez, and R. G. Harvey, *Carcinogenesis*, **4**, 519 (1983).
5. M. R. Osborne, P. Brookes, H. Lee, and R. G. Harvey, *Carcinogenesis*, **7**, 1345 (1986).
6. H. W. S. King, M. R. Osborne, and P. Brookes, *Int. J. Cancer* 1977, **20**, 564. *Ibid. Chem.-Biol. Interactions*, **20**, 367 (1978).
7. D. R. Thakker, W. Levin, A. W. Wood, A. H. Conney, T. A. Stoming, and D. M. Jerina, *J. Am. Chem. Soc.*, **100**, 645 (1978).
8. A. Eastman and E. Bresnick, *Cancer Res.*, **39**, 4316 (1979).
9. W. Levin, M. K. Buening, A. W. Wood, R. L. Chang, D. R. Thakker, D. M. Jerina, and A. H. Conney, *Cancer Res.*, **39**, 3549 (1979).
10. R. G. Harvey, C. Cortez, and S. A. Jacobs, *J. Org. Chem.*, **47**, 2120 (1982).
11. P. J. Garegg and B. Samuelsson, *Synthesis*, 469 (1978).
12. It is likely that methanol is formed preferentially as a consequence of the greater reactivity of the benzylic carbon atom.
13. L. B. Fieser and E. B. Hershberg, *J. Am. Chem. Soc.*, **60**, 2542 (1938).
14. P. Sims, *Biochem. J.* **98**, 215 (1966).

(Received June 15, 1987; in revised form August 6, 1987)